

Kinetic hard-modelling and spectral validation of rank-deficient spectroscopic data

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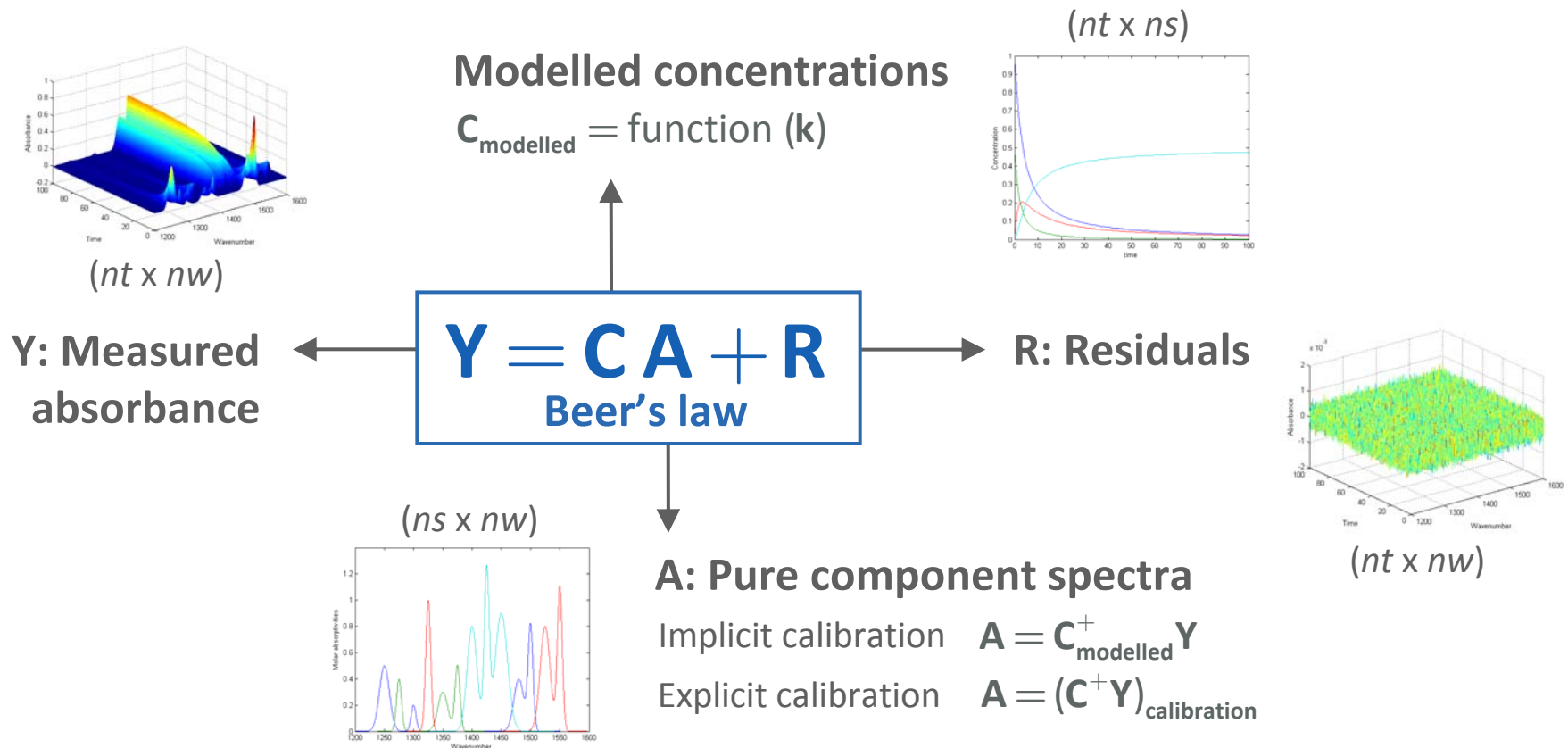
ETH Zürich, Institute for Chemical and Bioengineering,
Safety and Environmental Technology Group, Switzerland

Direct fitting by modelling concentrations

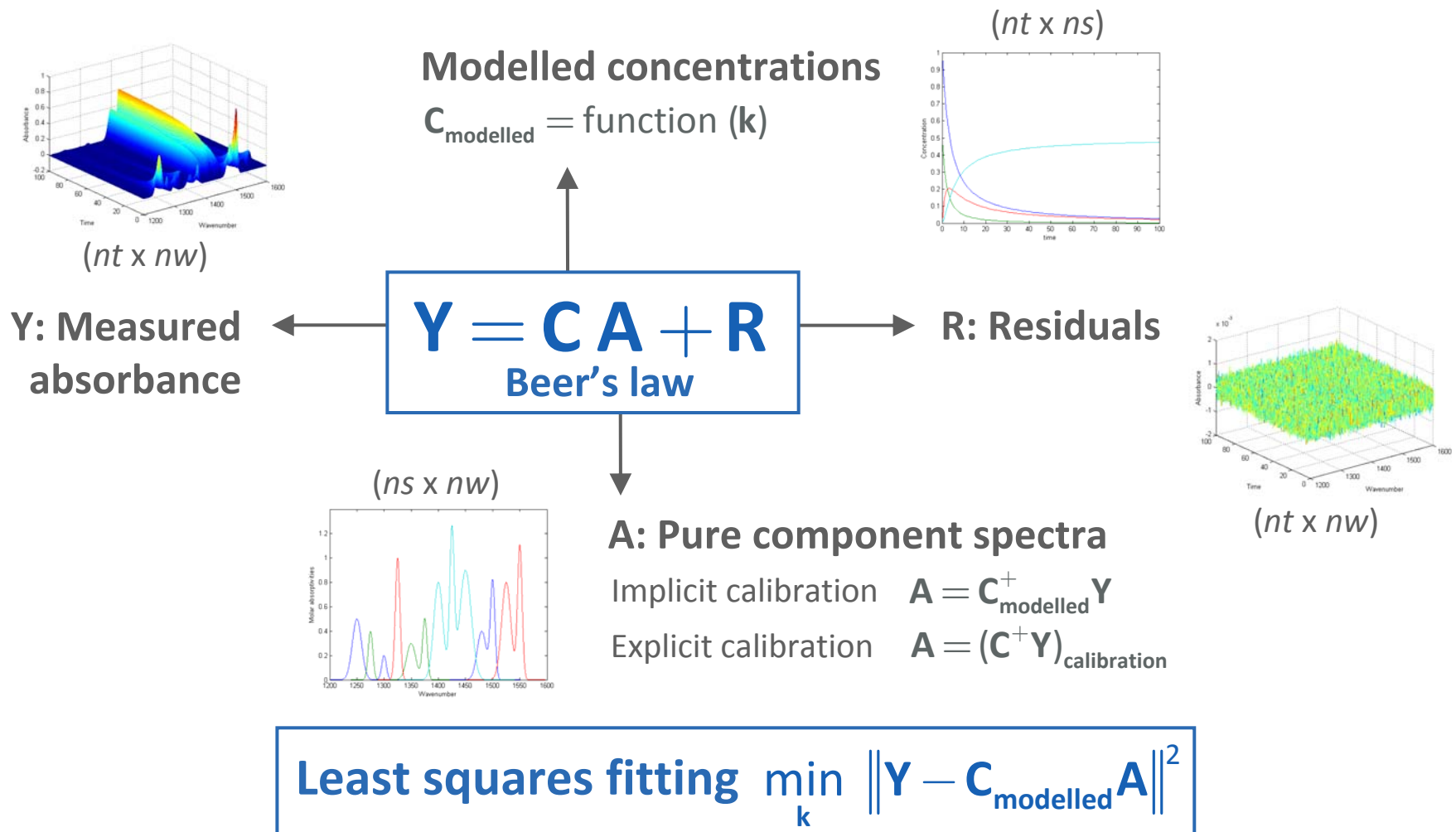
$$Y = CA + R$$

Beer's law

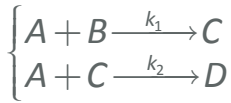
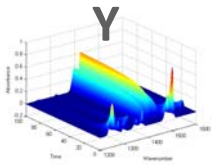
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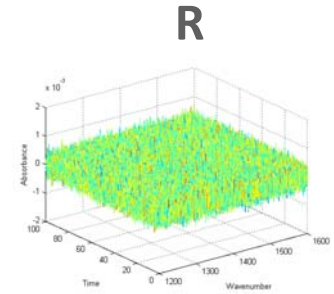
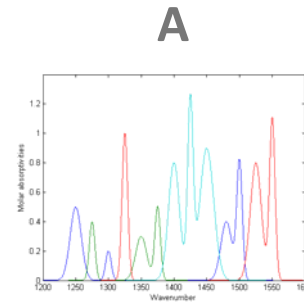
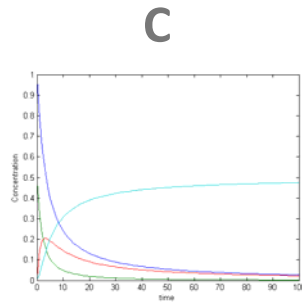
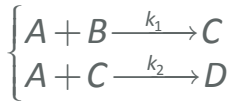
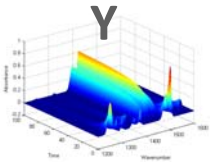


Rank deficiency in spectroscopy



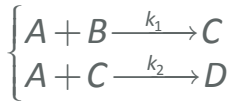
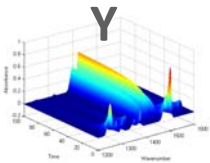
Rank deficiency in spectroscopy

Beer's law ($Y = C A$)
 $ns = 4$ species

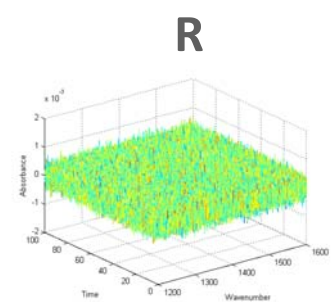
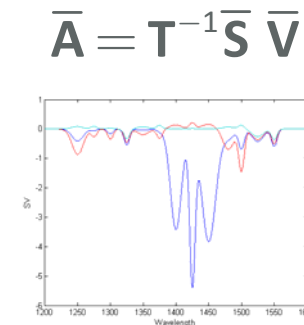
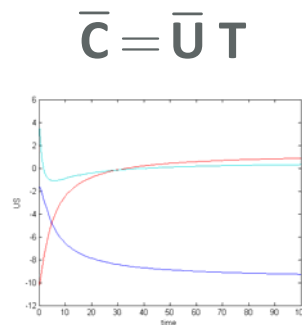
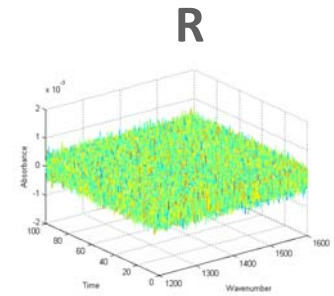
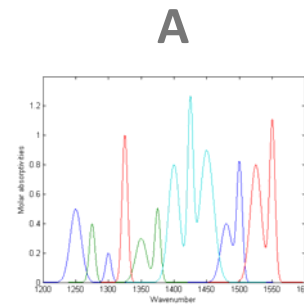
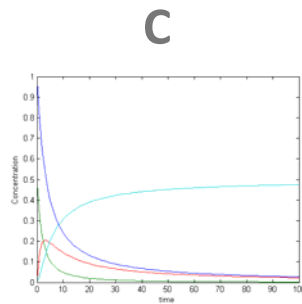


Rank deficiency in spectroscopy

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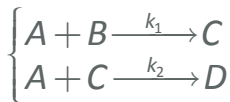
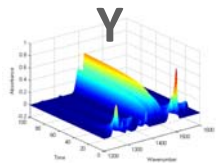
PCA / TFA ($\bar{Y} = \bar{U} \bar{S} \bar{V}$)
 $nc = 3$ factors



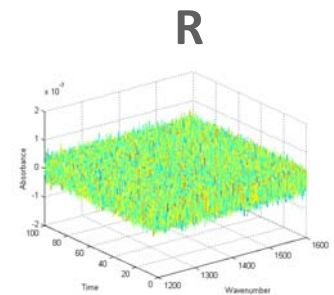
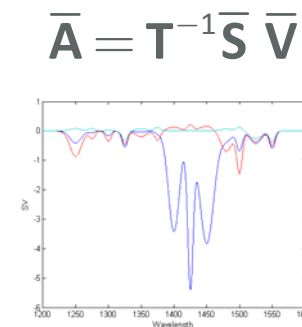
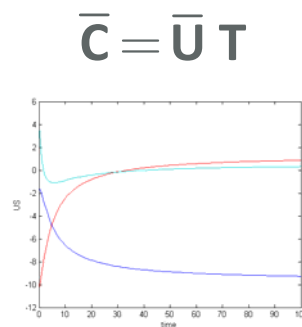
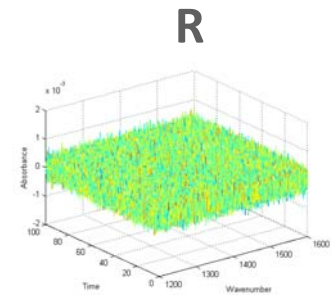
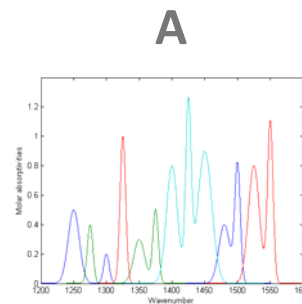
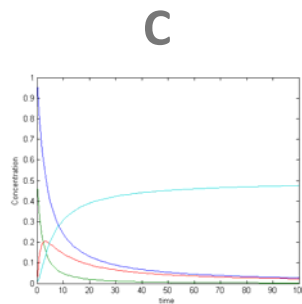
Where **T** is a transformation matrix of dimensions ($nc \times nc$)

Rank deficiency in spectroscopy

Beer's law ($\mathbf{Y} = \mathbf{C} \mathbf{A}$)
 $ns = 4$ species



PCA / TFA ($\bar{\mathbf{Y}} = \bar{\mathbf{U}} \bar{\mathbf{S}} \bar{\mathbf{V}}$)
 $nc = 3$ factors



Where \mathbf{T} is a transformation matrix of dimensions ($nc \times nc$)

Spectroscopic data matrix \mathbf{Y} is rank deficient when:

Significant factors (nc) in PCA < number of reactive species (ns) $\Leftrightarrow \text{rank}(\mathbf{Y}) < ns$

Sources and problems of rank deficiency

$$\mathbf{Y} = \mathbf{C} \mathbf{A}$$

Sources and problems of rank deficiency

$$\mathbf{Y} = \mathbf{C} \mathbf{A}$$

Rank deficiency in \mathbf{Y} is due to

Linear dependencies in \mathbf{C} and/or

Linear dependencies in \mathbf{A}

Mathematical ambiguity in
case of implicit calibration

\mathbf{A} cannot be computed by $\mathbf{C}^+\mathbf{Y}$
as \mathbf{A} is not unique

Not discussed here

All spectra in \mathbf{A} are assumed
to be linearly independent

Example: two species that are consumed
or generated at the same rate

$$\text{rank}(\mathbf{Y}) = \min[\text{rank}(\mathbf{C}), \text{rank}(\mathbf{A})] = \text{rank}(\mathbf{C})$$



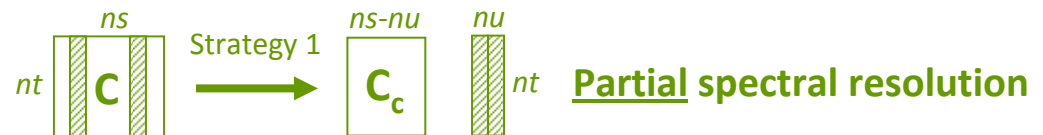
Strategies to treat rank deficiency in concentrations

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Model reduction

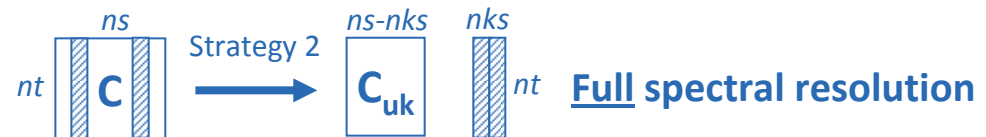
Strategy 1

define nu uncoloured species



Strategy 2

include nks known spectra in the analysis (explicit calibration)

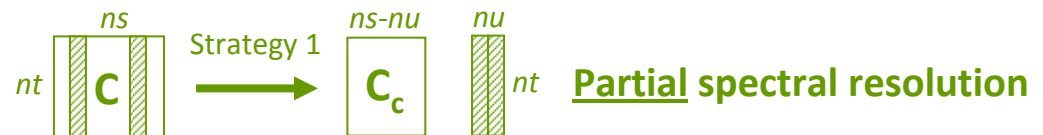


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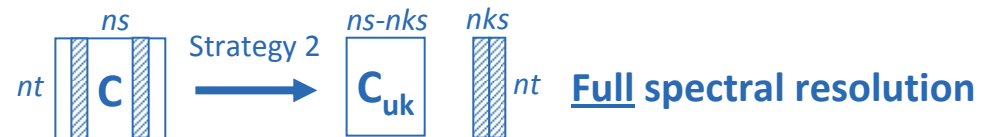
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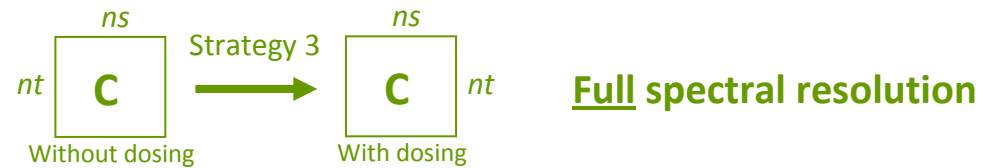
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Rank augmentation

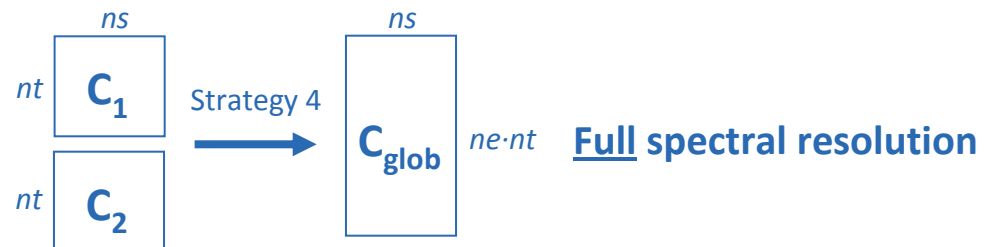
Strategy 3

dose one or more species in nf dosing steps



Strategy 4:

perform ne additional experiments by varying the initial concentrations (second order global analysis)

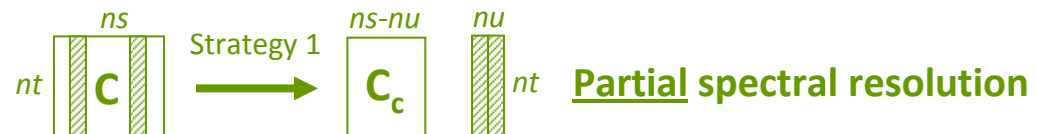


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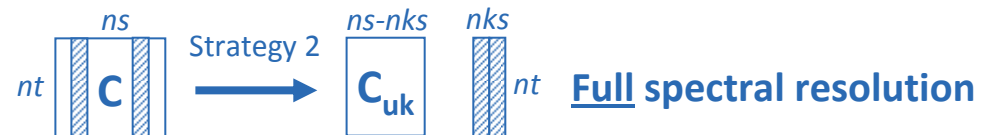
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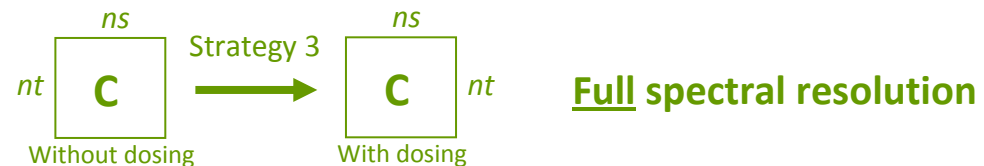
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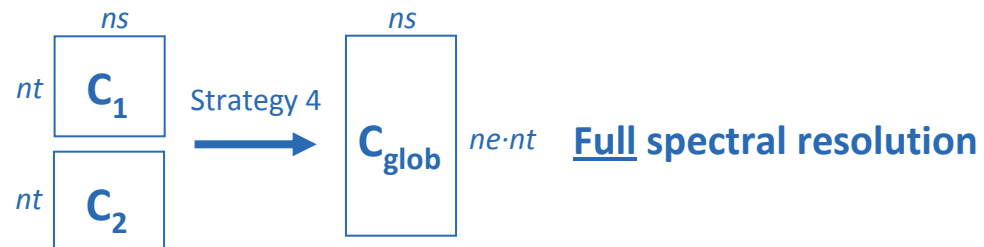
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How to identify the species to include in these four Strategies ?



Rank and kernel of the concentration matrix

Rank of \mathbf{C}

Number of linearly independent columns or rows in \mathbf{C}

>> Defines the maximum number of columns (species) to keep
in Strategy 1 ($ns - nu$) and Strategy 2 ($ns - nks$)

kernel of \mathbf{C}

Vector space spanned by the vectors forming
the null space $\mathbf{0}$ when multiplied by \mathbf{C}

$$\text{e.g. } \ker \mathbf{C} = \begin{bmatrix} 0.8 & 0.1 \\ -0.3 & -0.7 \\ 0 & 0 \\ 0.5 & -0.6 \end{bmatrix} \begin{matrix} A \\ B \\ C \\ D \end{matrix}$$

>> $\ker \mathbf{C}$ defines a mass balance equation: $\mathbf{C} (\ker \mathbf{C}) = \mathbf{0}$

>> $\ker \mathbf{C}$ defines which columns of \mathbf{C} are linearly dependent or independent

Interpretation of the kernel of \mathbf{C}

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Species without
zero rows in $\text{ker } \mathbf{C}$

Species with
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kernel of \mathbf{C}

Linearly dependent species

Linearly independent species

Consequences for Strategies 1 – 4 ?

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Consequences for Strategies 1 – 4

Strategy 1

Define this species as uncoloured...

Do not define this species as uncoloured...

Strategy 2

Include its pure spectrum...

Do not provide its pure spectrum...

Strategy 3

Dose this species...

Do not dose this species...

Strategy 4

Vary its initial concentration...

Do not vary its initial concentration...

... to break/avoid rank deficiency in \mathbf{C}

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Can we obtain this information without numerical integration of the rate laws, i.e. using a time invariant approach?



Modelling kinetic concentration profiles

Kinetic rate law

$$\dot{\mathbf{x}}_t = \prod_{\text{along columns}} (\mathbf{c}_t^T \mathbf{1})^{\mathbf{E}^T} \text{DIAG}(\mathbf{k})$$

Reactant coefficients
 $\mathbf{E} \ (nr \times ns)$

Rate constants
 $\mathbf{k} \ (1 \times nr)$

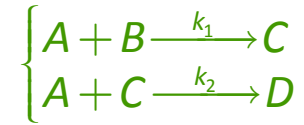
Stoichiometric coefficients
 $\mathbf{N} \ (nr \times ns)$

Concentration profiles

$$\dot{\mathbf{c}}_t = \dot{\mathbf{x}}_t \mathbf{N} + f_{in,t} v_t^{-1} (\mathbf{c}_{in,t} - \mathbf{c}_t) - f_{out,t} v_t^{-1} \mathbf{c}_t$$

This system of ODE is numerically integrated with initial concentrations $\mathbf{c}_0 \ (1 \times ns)$ and results in the concentration matrix $\mathbf{C} \ (nt \times ns)$

Modelling kinetic concentration profiles



Kinetic rate law

$$\dot{\mathbf{x}}_t = \prod_{\text{along columns}} (\mathbf{c}_t^T \mathbf{1})^{\mathbf{E}^T} \text{DIAG}(\mathbf{k}) \begin{bmatrix} k_1 c_{t,A} c_{t,B} & k_2 c_{t,A} c_{t,C} \end{bmatrix}$$

$$\begin{bmatrix} 1 & 1 & 0 & 0 \\ 1 & 0 & 1 & 0 \end{bmatrix}$$

Reactant coefficients
 $\mathbf{E} (nr \times ns)$

Rate constants
 $\mathbf{k} (1 \times nr) \quad [k_1 \ k_2]$

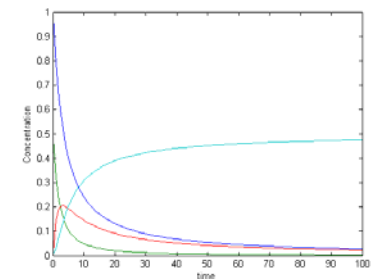
$$\begin{bmatrix} -1 & -1 & 1 & 0 \\ -1 & 0 & -1 & 0 \end{bmatrix}$$

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A time invariant matrix Ω equivalent to C

$$\Omega = \begin{bmatrix} \frac{(\mu \mathbf{1})^{\bullet E^T} \text{DIAG}(\mathbf{k}) \mathbf{N}}{\mathbf{c}_0} \\ \mathbf{c}_{\text{in}} \\ \mathbf{c}_0^{\text{ne}} \end{bmatrix} \quad (ns + 1 + nf + ne \times ns - nu - nks) \quad \ker \Omega = (\ker C) \mathbf{T}_{\text{lin}}$$

μ an arbitrary positive scalar
different from 0 and 1

$\mathbf{1}$ ($ns \times nr$) matrix comprised of ones

\mathbf{E} ($nr \times ns$) matrix of reactant coefficients

$\bullet \mathbf{E}^T$ element-wise raise to the power of \mathbf{E}^T

DIAG operator generating a diagonal
matrix from a vector argument

\mathbf{k} ($1 \times nr$)

\mathbf{N} ($nr \times ns$)

\mathbf{c}_0 ($1 \times ns$)

\mathbf{c}_{in} ($nf \times ns$)

\mathbf{c}_0^{ne} ($ne \times ns$)

vector of rate constants

matrix of stoichiometric coefficients

vector of initial concentrations

matrix of the dosing concentrations
corresponding to the nf dosing steps

matrix of the varied initial concentrations
corresponding to the ne additional experiments

Advantages of the time invariant approach:

No numerical integration required

Analytical (symbolic) relationship between the experimental conditions (\mathbf{c}_0 , \mathbf{c}_{in} , \mathbf{c}_0^{ne})

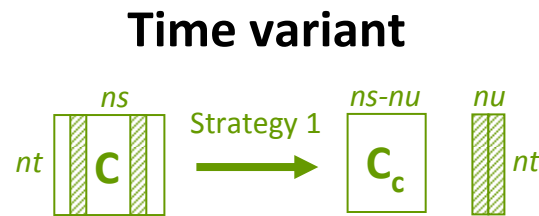
Mathematical description and proof: Billeter et al, Chemom. Intell. Lab. Syst., 95 (2009), 170

Strategies to treat rank deficiency in concentrations

Model reduction

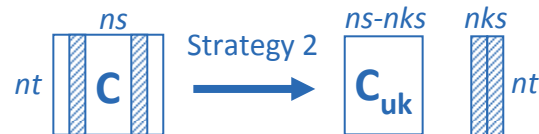
Strategy 1

define nu uncoloured species



Strategy 2

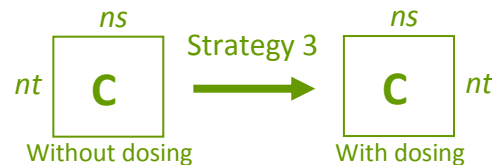
include nks known spectra in the analysis (explicit calibration)



Rank augmentation

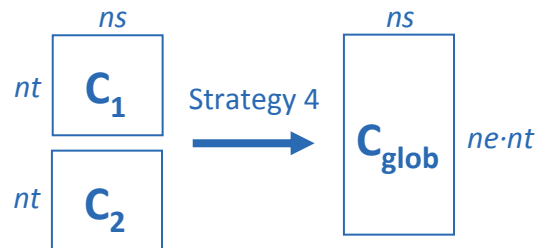
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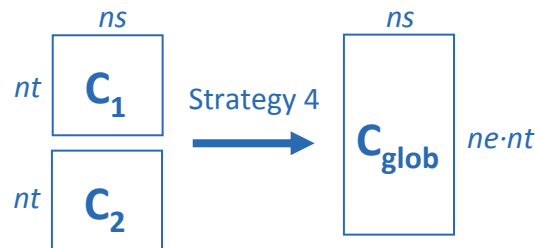
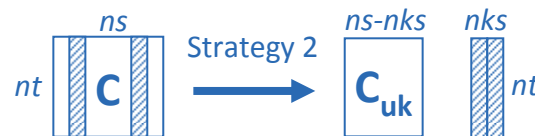
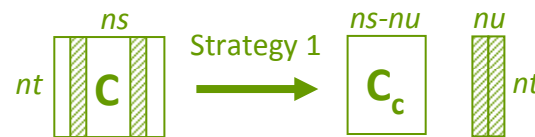
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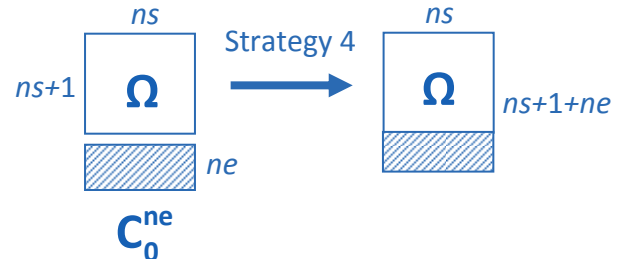
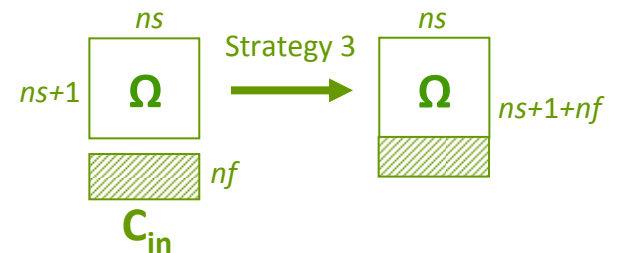
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Time variant



Time invariant



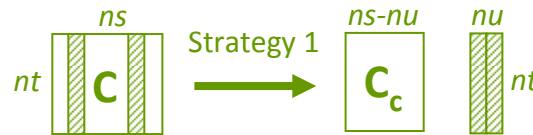
Spectral consequence of Strategy 1

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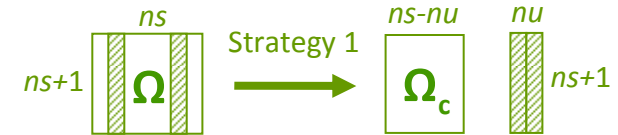
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define nu uncoloured species

Time variant



Time invariant



Spectral consequence of Strategy 1

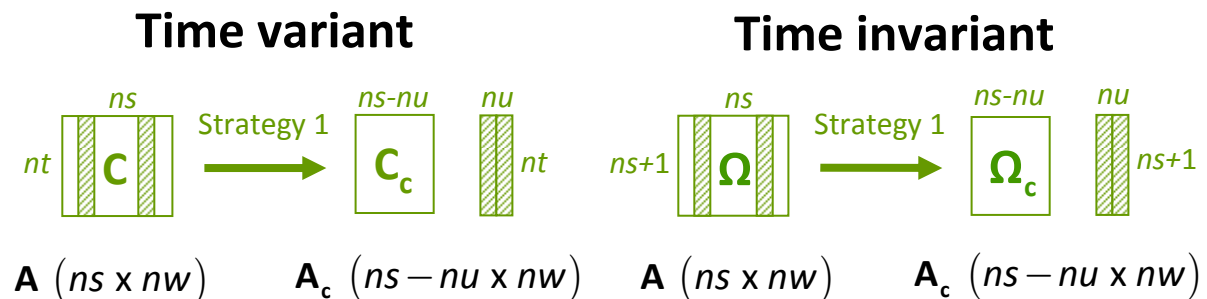
Model reduction

Strategy 1

define nu uncoloured species

Spectral contributions of the nu uncoloured species are linearly transferred into the fitted pure spectra of the coloured species

The fitted component spectra \mathbf{A}_c of the $(ns - nu)$ coloured species are comprised of linear combinations of the ns true pure component spectra \mathbf{A}



Spectral consequence of Strategy 1

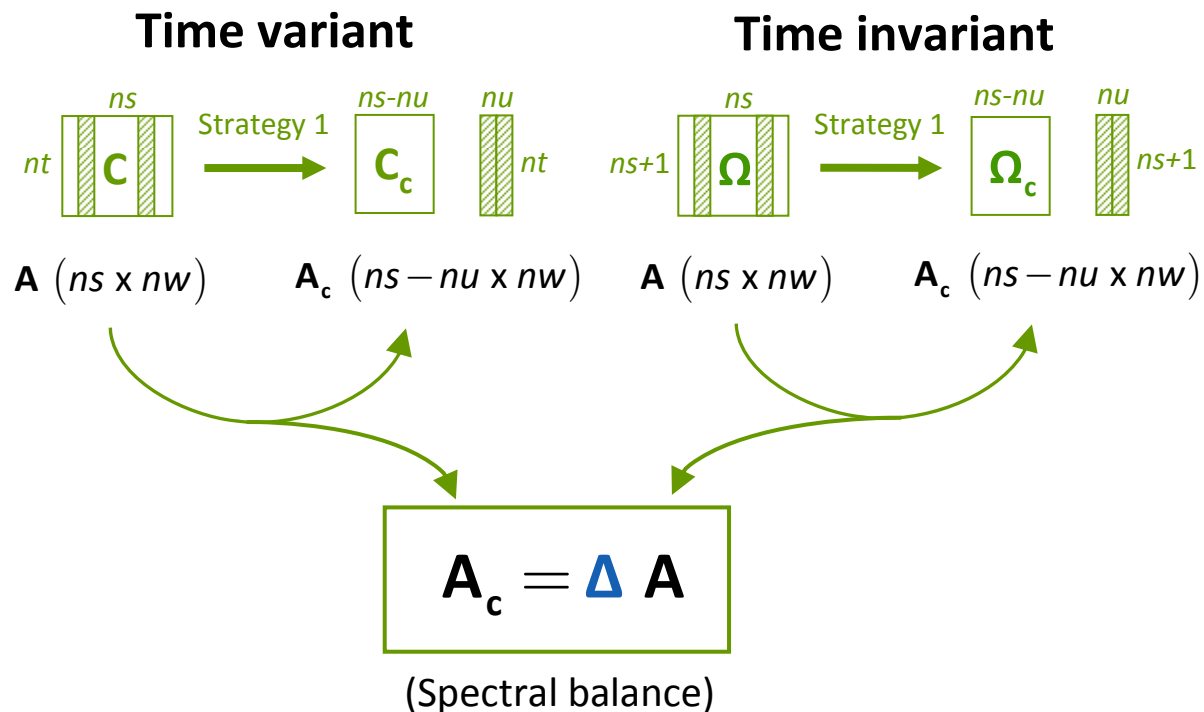
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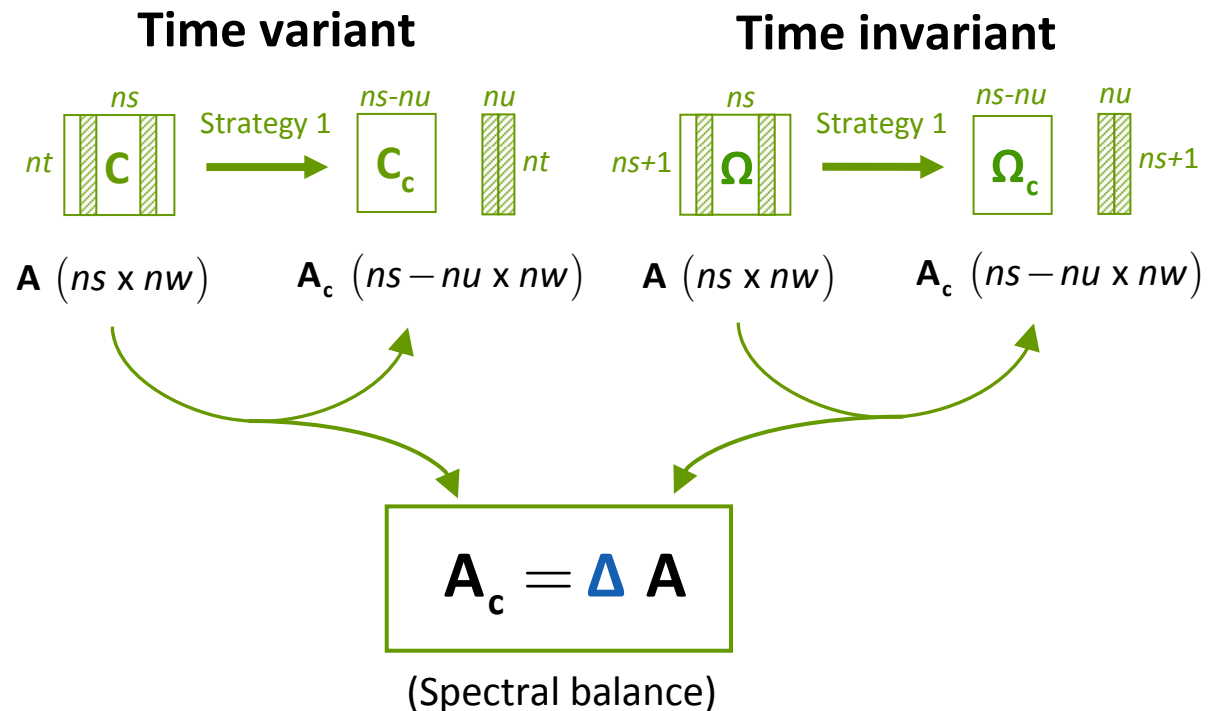
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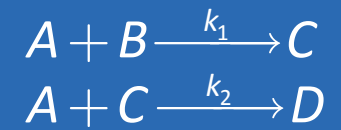
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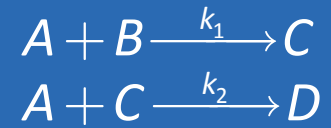


$$\Delta (ns - nu \times ns) = C_c^+ C = \left(\Omega \Big|_{\text{comprised of coloured species}} \right)^+ \Omega \Big|_{\text{comprised of all species}}$$

Example: Calculation of the kernel



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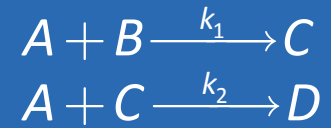
$$\Omega = \left[\frac{(\mu \mathbf{1})^{\mathbf{E}^T} \text{DIAG}(\mathbf{k}) \mathbf{N}}{\mathbf{c}_0} \right] = \begin{array}{c|cccc} & A & B & C & D \\ \hline -\mu k_1 - \mu k_2 & -\mu k_1 & \mu k_1 - \mu k_2 & \mu k_2 \\ -\mu k_1 - k_2 & -\mu k_1 & \mu k_1 - k_2 & k_2 \\ -k_1 - \mu k_2 & -k_1 & k_1 - \mu k_2 & \mu k_2 \\ -k_1 - k_2 & -k_1 & k_1 - k_2 & k_2 \\ \hline c_{0,A} & \alpha c_{0,A} & 0 & 0 \end{array}$$

Matlab code (8 lines)

```
>> syms c0A alpha mu k1 k2
>> N      = [-1 -1 1 0; -1 0 -1 1];
>> E      = [1 1 0 0; 1 0 1 0];
>> k      = [k1, k2];
>> c0     = [c0A, alpha*c0A, 0, 0];
>> one    = ones(size(E'));
>> omega  = [(mu*one).^(E')*diag(k)*N; c0];
>> null(omega)
ans =
    -alpha
         1
    1-alpha
    1-2*alpha
```

$$\alpha = \frac{c_{0,B}}{c_{0,A}}$$

Example: Calculation of the kernel



$$\Omega = \left[\frac{(\mu \mathbf{1})^{\bullet E^T} \text{DIAG}(\mathbf{k}) \mathbf{N}}{\mathbf{c}_0} \right] = \begin{array}{c|cccc} & A & B & C & D \\ \hline -\mu k_1 - \mu k_2 & -\mu k_1 & \mu k_1 - \mu k_2 & \mu k_2 \\ -\mu k_1 - k_2 & -\mu k_1 & \mu k_1 - k_2 & k_2 \\ -k_1 - \mu k_2 & -k_1 & k_1 - \mu k_2 & \mu k_2 \\ -k_1 - k_2 & -k_1 & k_1 - k_2 & k_2 \\ \hline c_{0,A} & \alpha c_{0,A} & 0 & 0 \end{array}$$

$$\ker \Omega = \begin{array}{c|c} \begin{bmatrix} -\alpha \\ 1 \\ 1 - \alpha \\ 1 - 2\alpha \end{bmatrix} & \begin{array}{l} A \\ B \\ C \\ D \end{array} \end{array}$$

Matlab code (8 lines)

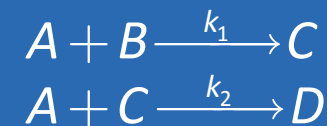
```
>> syms c0A alpha mu k1 k2
>> N      = [-1 -1 1 0; -1 0 -1 1];
>> E      = [1 1 0 0; 1 0 1 0];
>> k      = [k1, k2];
>> c0     = [c0A, alpha*c0A, 0, 0];
>> one     = ones(size(E'));
>> omega   = [(mu*one).^(E')*diag(k)*N; c0];
>> null(omega)
```

ans =

```
-alpha
      1
1-alpha
1-2*alpha
```

$$\alpha = \frac{c_{0,B}}{c_{0,A}}$$

Example: Calculation of the kernel



$$\Omega = \left[\frac{(\mu \mathbf{1})^{\text{E}^T} \text{DIAG}(\mathbf{k}) \mathbf{N}}{\mathbf{c}_0} \right] = \begin{array}{c} \begin{array}{cccc} & A & B & C & D \\ \begin{array}{l} -\mu k_1 - \mu k_2 \\ -\mu k_1 - k_2 \\ -k_1 - \mu k_2 \\ -k_1 - k_2 \end{array} & \begin{array}{l} -\mu k_1 \\ -\mu k_1 \\ -k_1 \\ -k_1 \end{array} & \begin{array}{l} \mu k_1 - \mu k_2 \\ \mu k_1 - k_2 \\ k_1 - \mu k_2 \\ k_1 - k_2 \end{array} & \begin{array}{l} \mu k_2 \\ k_2 \\ \mu k_2 \\ k_2 \end{array} \\ \hline c_{0,A} & \alpha c_{0,A} & 0 & 0 \end{array} \end{array}$$

$$\ker \Omega = \begin{array}{c} \begin{bmatrix} -\alpha \\ 1 \\ 1 - \alpha \\ 1 - 2\alpha \end{bmatrix} \begin{array}{l} A \\ B \\ C \\ D \end{array} \end{array}$$

Determination of the roots of the kernel

If $\alpha = 1$, species C is linearly independent from the others

If $\alpha = 0.5$, species D is linearly independent from the others

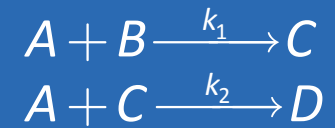
If $\alpha \neq 1$ or 0.5, all species are linearly dependent from the others

Matlab code (8 lines)

```
>> syms c0A alpha mu k1 k2
>> N = [-1 -1 1 0; -1 0 -1 1];
>> E = [1 1 0 0; 1 0 1 0];
>> k = [k1, k2];
>> c0 = [c0A, alpha*c0A, 0, 0];
>> one = ones(size(E'));
>> omega = [(mu*one).^(E')*diag(k)*N; c0];
>> null(omega)
ans =
    -alpha
         1
    1-alpha
    1-2*alpha
```

$$\alpha = \frac{c_{0,B}}{c_{0,A}}$$

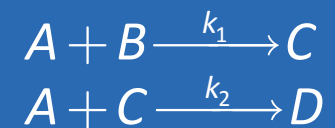
Example: Design of experiments



$$\ker \mathbf{\Omega} \ (4 \times 1) = \begin{bmatrix} \overset{A}{-\alpha} & \overset{B}{1} & \overset{C}{1-\alpha} & \overset{D}{1-2\alpha} \end{bmatrix}^T$$

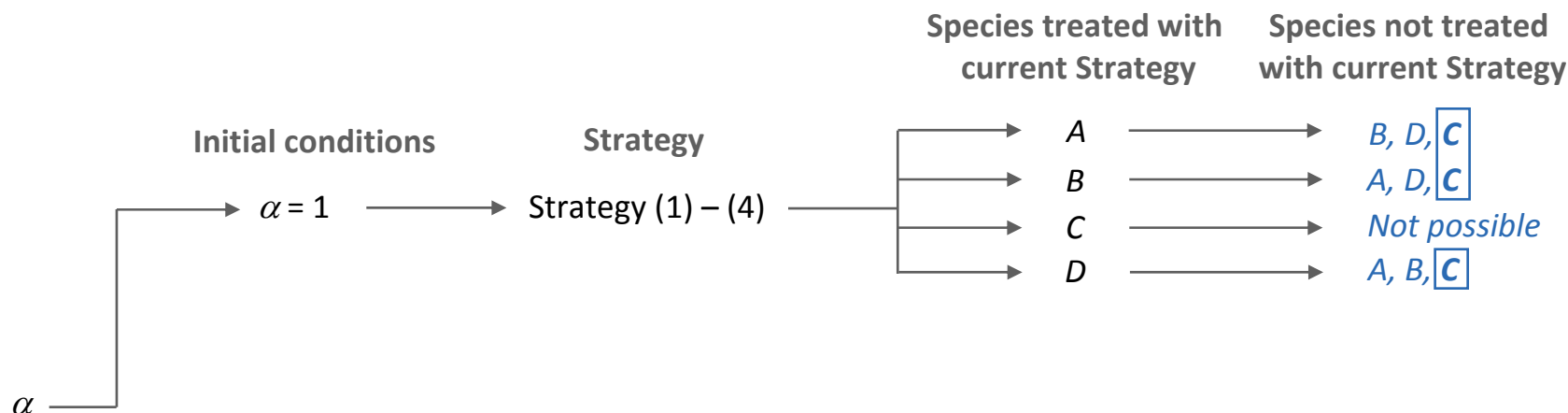
$\dim(\ker \mathbf{\Omega}) = 1$, i.e. only one species
has to be considered in Strategies (1) – (4)

Example: Design of experiments

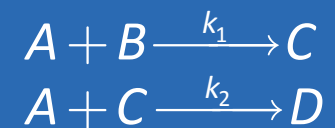


$$\ker \mathbf{\Omega} \ (4 \times 1) = \begin{bmatrix} A & B & C & D \\ -\alpha & 1 & 1-\alpha & 1-2\alpha \end{bmatrix}^T$$

$\dim(\ker \mathbf{\Omega}) = 1$, i.e. only one species has to be considered in Strategies (1) – (4)

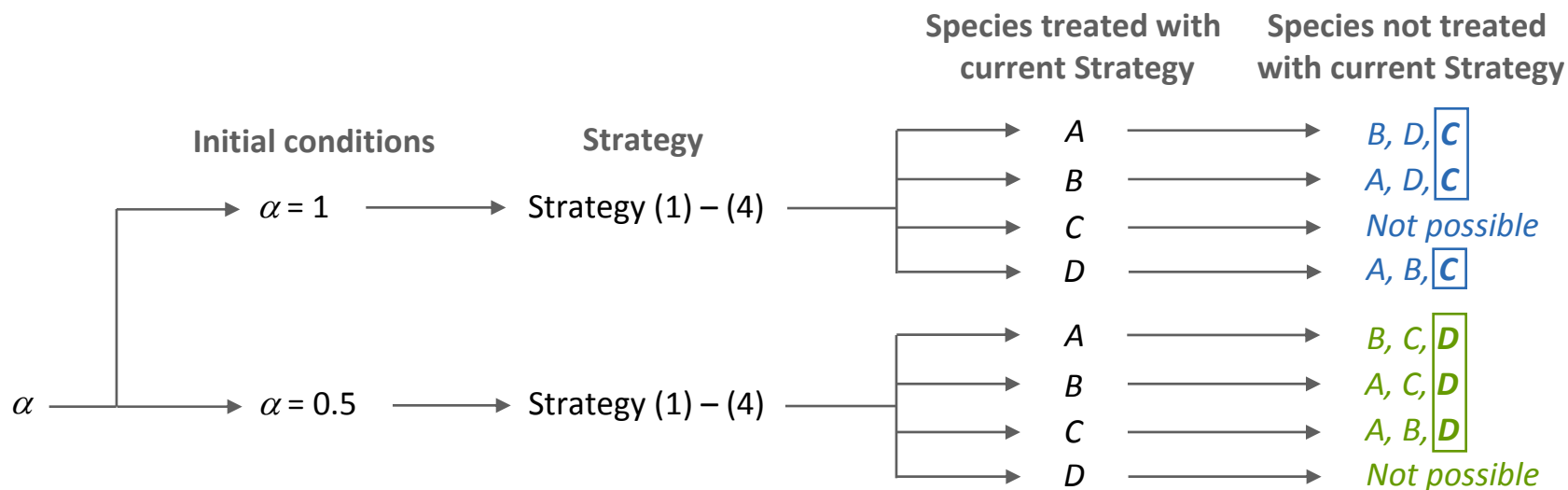


Example: Design of experiments

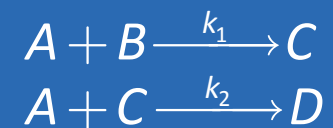


$$\ker \mathbf{\Omega} \ (4 \times 1) = \begin{bmatrix} A & B & C & D \\ -\alpha & 1 & 1-\alpha & 1-2\alpha \end{bmatrix}^T$$

$\dim(\ker \mathbf{\Omega}) = 1$, i.e. only one species has to be considered in Strategies (1) – (4)

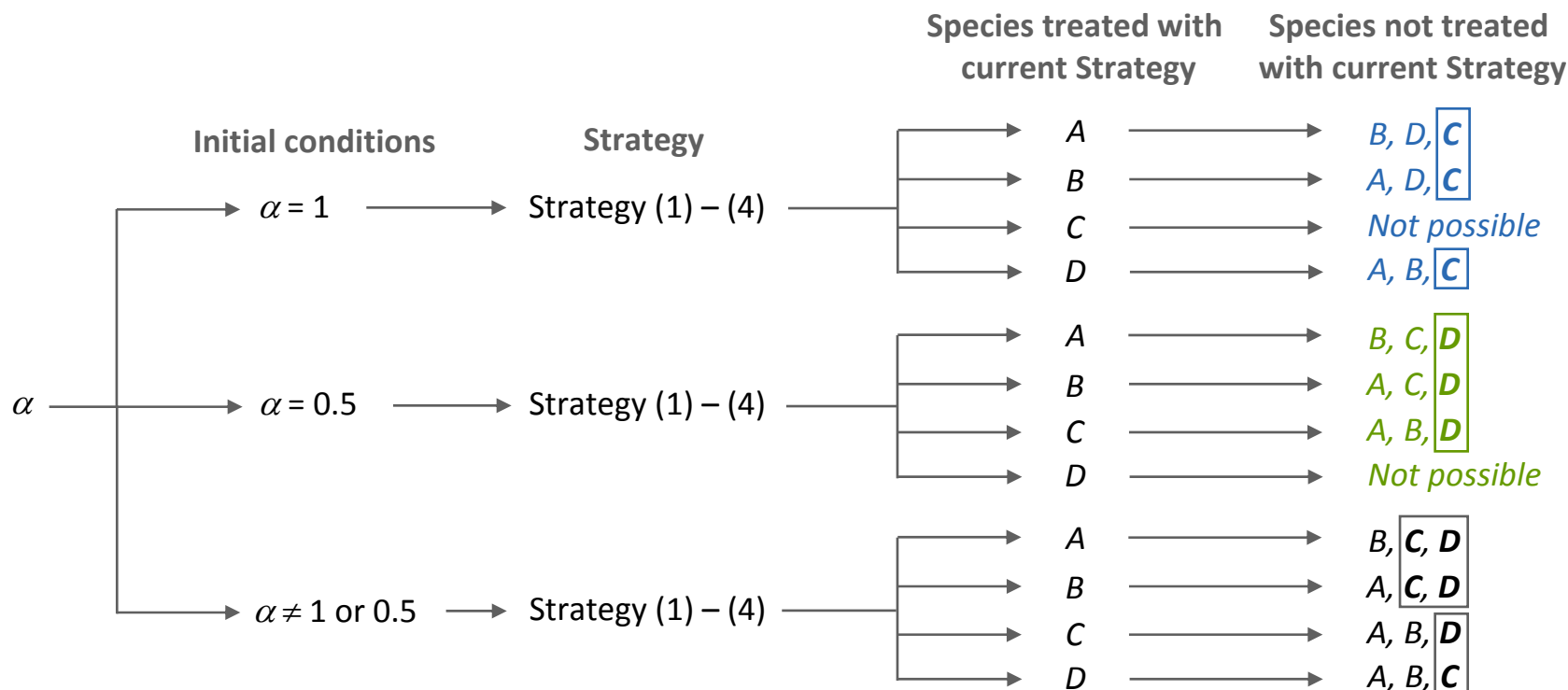


Example: Design of experiments

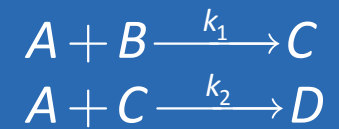


$$\ker \mathbf{\Omega} \ (4 \times 1) = \begin{bmatrix} A & B & C & D \\ -\alpha & 1 & 1-\alpha & 1-2\alpha \end{bmatrix}^T$$

$\dim(\ker \mathbf{\Omega}) = 1$, i.e. only one species has to be considered in Strategies (1) – (4)



Example: Spectral consequence of Strategy 1



Model reduction

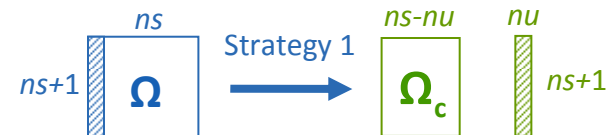
Strategy 1

define nu uncoloured species

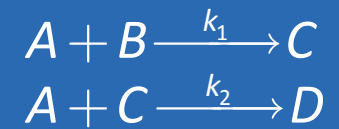
Time variant



Time invariant



Example: Spectral consequence of Strategy 1



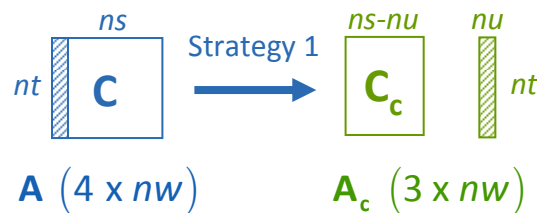
Model reduction

Strategy 1

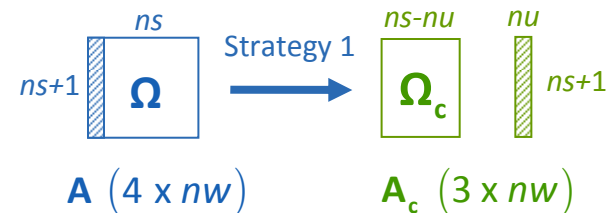
define nu uncoloured species

Species A uncoloured

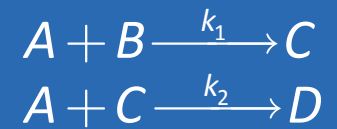
Time variant



Time invariant



Example: Spectral consequence of Strategy 1



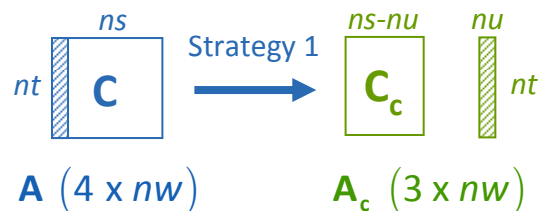
Model reduction

Strategy 1

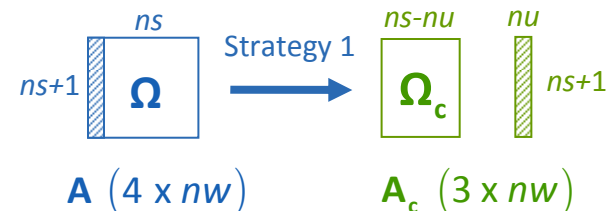
define nu uncoloured species

Species A uncoloured

Time variant



Time invariant

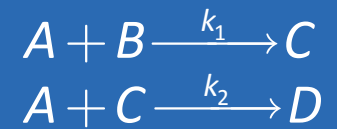


$$\Omega_{\text{comprised of all species}} = \begin{array}{c|cccc} & A & B & C & D \\ \hline A & -\mu k_1 - \mu k_2 & -\mu k_1 & \mu k_1 - \mu k_2 & \mu k_2 \\ B & -\mu k_1 - k_2 & -\mu k_1 & \mu k_1 - k_2 & k_2 \\ C & -k_1 - \mu k_2 & -k_1 & k_1 - \mu k_2 & \mu k_2 \\ D & -k_1 - k_2 & -k_1 & k_1 - k_2 & k_2 \\ \hline & c_{0,A} & \alpha c_{0,A} & 0 & 0 \end{array}$$

$$\Omega_{\text{comprised of coloured species}} = \begin{array}{c|ccc} & 'B' & 'C' & 'D' \\ \hline 'B' & -\mu k_1 & \mu k_1 - \mu k_2 & \mu k_2 \\ 'C' & -\mu k_1 & \mu k_1 - k_2 & k_2 \\ 'D' & -k_1 & k_1 - \mu k_2 & \mu k_2 \\ & -k_1 & k_1 - k_2 & k_2 \\ \hline & \alpha c_{0,A} & 0 & 0 \end{array}$$

$$\begin{bmatrix} a_{C'B'} \\ a_{C'C'} \\ a_{C'D'} \end{bmatrix} = \Delta A = (CC^+)A = \left(\left(\Omega_{\text{comprised of coloured species}} \right)^+ \Omega_{\text{comprised of all species}} \right) A = \begin{bmatrix} \alpha^{-1} & 1 & 0 & 0 \\ \alpha^{-1} - 1 & 0 & 1 & 0 \\ \alpha^{-1} - 2 & 0 & 0 & 1 \end{bmatrix} \begin{bmatrix} a_{A,:} \\ a_{B,:} \\ a_{C,:} \\ a_{D,:} \end{bmatrix}$$

Example: Spectral consequence of Strategy 1



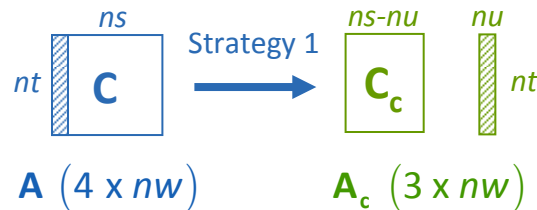
Model reduction

Strategy 1

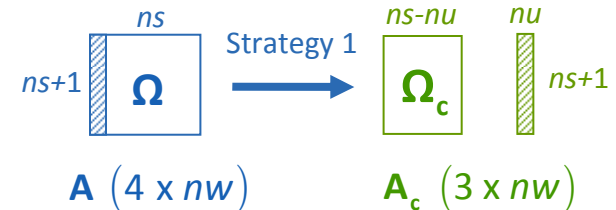
define nu uncoloured species

Species A uncoloured

Time variant



Time invariant



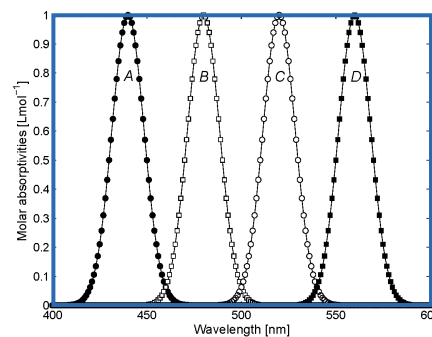
All species

$$\mathbf{\Omega}_{\text{comprised of all species}} = \begin{bmatrix} \begin{matrix} A & B & C & D \end{matrix} \\ \begin{matrix} -\mu k_1 - \mu k_2 & -\mu k_1 & \mu k_1 - \mu k_2 & \mu k_2 \\ -\mu k_1 - k_2 & -\mu k_1 & \mu k_1 - k_2 & k_2 \\ -k_1 - \mu k_2 & -k_1 & k_1 - \mu k_2 & \mu k_2 \\ -k_1 - k_2 & -k_1 & k_1 - k_2 & k_2 \end{matrix} \\ \hline \begin{matrix} C_{0,A} & \alpha C_{0,A} & 0 & 0 \end{matrix} \end{bmatrix}$$

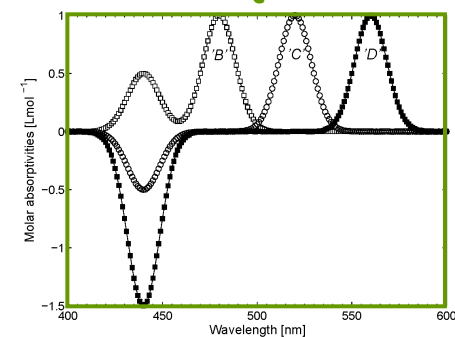
Coloured species

$$\mathbf{\Omega}_{\text{comprised of coloured species}} = \begin{bmatrix} \begin{matrix} 'B' & 'C' & 'D' \end{matrix} \\ \begin{matrix} -\mu k_1 & \mu k_1 - \mu k_2 & \mu k_2 \\ -\mu k_1 & \mu k_1 - k_2 & k_2 \\ -k_1 & k_1 - \mu k_2 & \mu k_2 \\ -k_1 & k_1 - k_2 & k_2 \end{matrix} \\ \hline \begin{matrix} \alpha C_{0,A} & 0 & 0 \end{matrix} \end{bmatrix}$$

True: \mathbf{A} ($4 \times nw$)



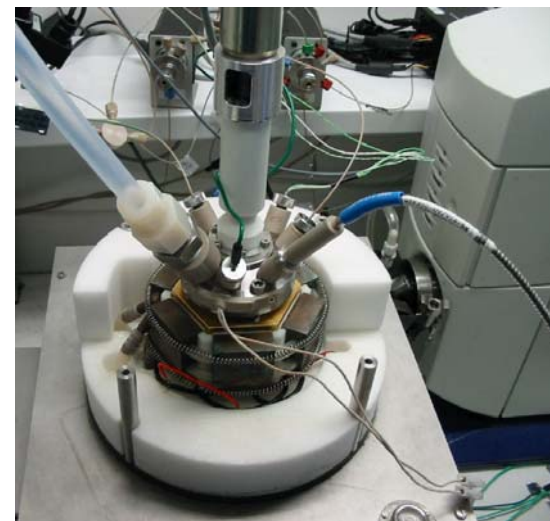
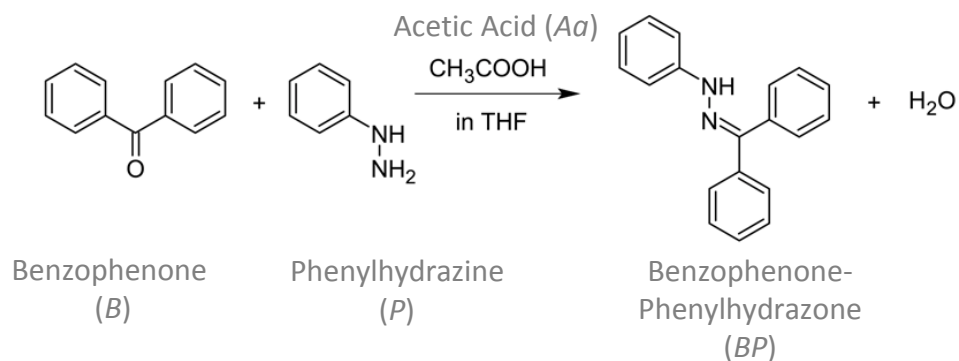
Fitted: \mathbf{A}_c ($3 \times nw$)



$$\begin{bmatrix} \mathbf{a}_{c, 'B';} \\ \mathbf{a}_{c, 'C';} \\ \mathbf{a}_{c, 'D';} \end{bmatrix} = \Delta \mathbf{A} = (\mathbf{C}\mathbf{C}^+) \mathbf{A} = \left(\left(\mathbf{\Omega}_{\text{comprised of coloured species}} \right)^+ \mathbf{\Omega}_{\text{comprised of all species}} \right) \mathbf{A} = \begin{bmatrix} \alpha^{-1} & 1 & 0 & 0 \\ \alpha^{-1} - 1 & 0 & 1 & 0 \\ \alpha^{-1} - 2 & 0 & 0 & 1 \end{bmatrix} \begin{bmatrix} \mathbf{a}_{A,:} \\ \mathbf{a}_{B,:} \\ \mathbf{a}_{C,:} \\ \mathbf{a}_{D,:} \end{bmatrix} = \begin{bmatrix} 0.5 \\ -0.5 \\ -1.5 \end{bmatrix} \mathbf{a}_{A,:} + \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix} \mathbf{a}_{B,:} + \begin{bmatrix} 0 \\ 1 \\ 0 \end{bmatrix} \mathbf{a}_{C,:} + \begin{bmatrix} 0 \\ 0 \\ 1 \end{bmatrix} \mathbf{a}_{D,:} \quad \text{when } \alpha = 2$$

An experimental case study

Overall reaction



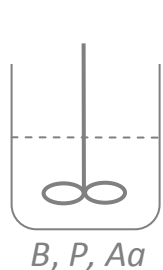
Kinetic mechanism



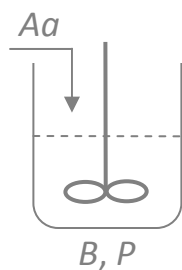
Reactor: CRC.v4 with FT-IR and UV-vis

Experimental conditions

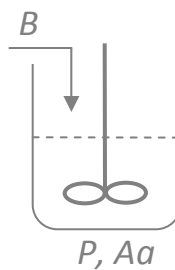
25°C, mid-IR (1200–1650 cm⁻¹), UV-vis (240–400 nm)



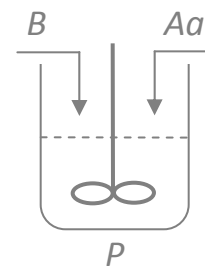
Batch conditions



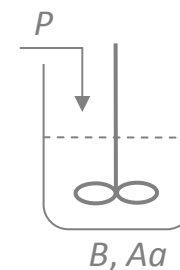
Dosing Aa



Dosing B

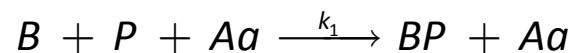


Dosing B + Aa

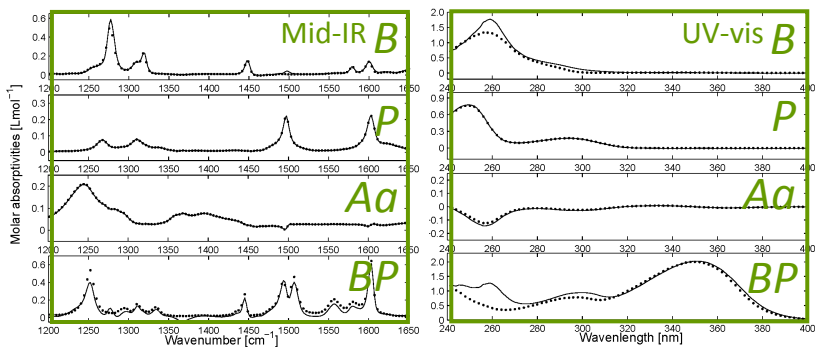


Dosing P

Spectral validation



Strategy (3): dosing



Species *B* and *Aa* dosed

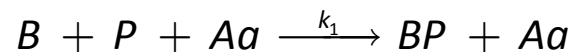
Full spectral resolution

— Fitted spectra

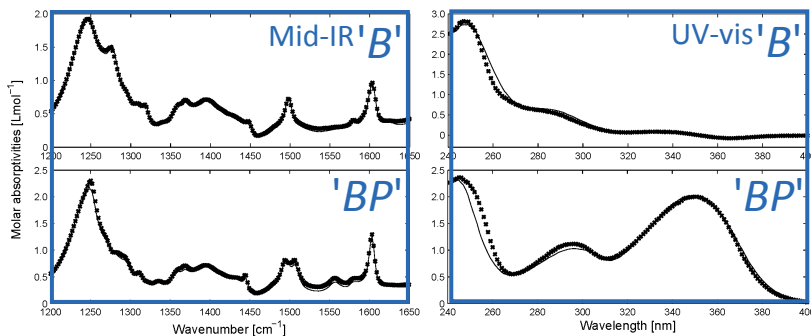
●●● Measured spectra

xxx Predicted spectra

Spectral validation



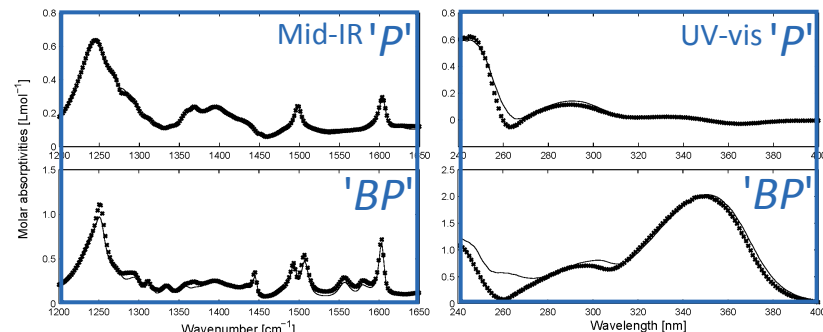
Strategy (1): uncoloured species



Species *P* and *Aa* set uncoloured
Partial spectral resolution

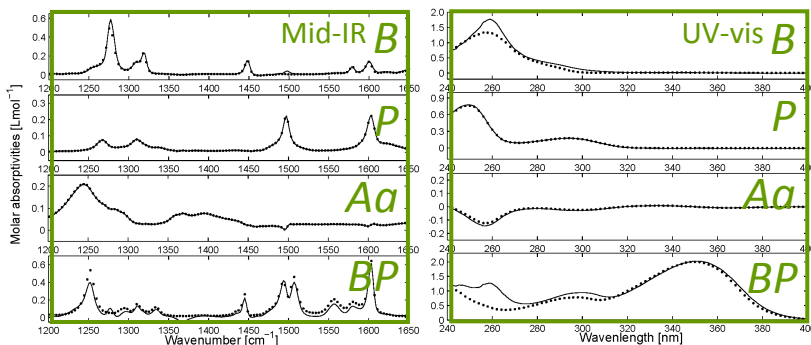
The model can now be validated
with and **without** rank deficiency!

Strategy (1)+(2): provided known spectrum



Pure spectrum of *B* provided
Aa set uncoloured
Partial spectral resolution

Strategy (3): dosing



Species *B* and *Aa* dosed
Full spectral resolution

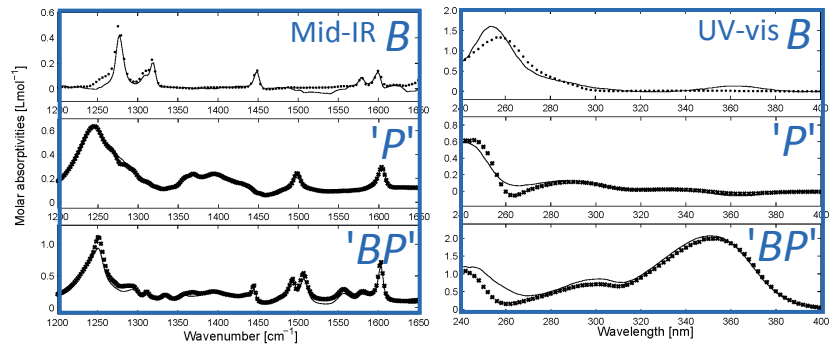
— Fitted spectra

●●● Measured spectra

xxx Predicted spectra

Partial spectral resolution

Strategy (1)+(4): second order global analysis



Initial concentration of *B* varied, *Aa* set uncoloured

Conclusions

Ker C

Linear dependencies, leading to rank deficiency in concentration matrix, can be elucidated using the **concept of kernel**

$$C(\text{ker } C) = 0$$

This equation defines a **mass balance equation** with time-invariant coefficients

Model reduction / rank augmentation Strategies

Species to be included in Strategies 1 – 4 can be identified by selecting the species with **non zero rows in the kernel**

Matrix Ω

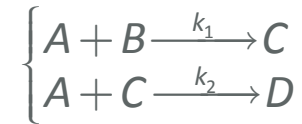
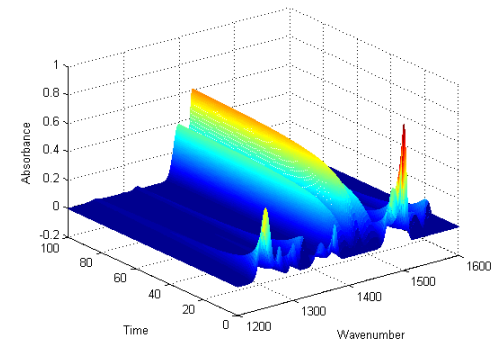
The kernel of the time-variant concentrations can be calculated without numerical integration using a simple **time-invariant approach**

Matrix Δ

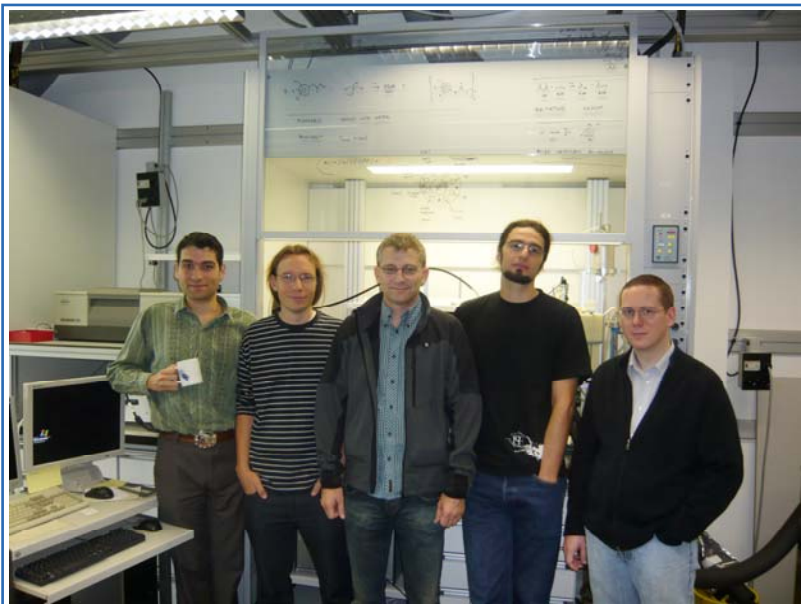
Linear combinations, observed in the fitted pure spectra when Strategy 1 (defining uncoloured species) is used, can be explained by a **spectral balance equation**

Spectral validation of kinetic models

Spectral validation of kinetic models is now possible even in case of **rank deficient spectroscopic data**



Reaction Analysis at ETH Zürich



Reaction Analysis Group
Leader: Dr. Bobby Neuhold

Dr. Bobby Neuhold, Chemist
Chemometrics, Kinetic hard-modelling
and Spectroscopy

Dr. Gilles Richner, Chemical Engineer
Calorimetry and Hydrogenation

Julien Billeter, Chemical Engineer
Chemometrics, Kinetic hard-modelling
and Spectroscopy

Tamás Godány, Chemical Engineer
Low-temperature alkylations

Sébastien Cap, Chemist
Particulate systems and Dissolution



Safety and
Environmental
Technology Group

under supervision of **Prof. K. Hungerbühler**

A blue-tinted banner image at the top of the slide showing a building with a large dome, likely a part of the ETH Zurich campus, with mountains in the background.

Thank you for your attention

Publications

Systematic prediction of linear dependencies in the concentration profiles
and implications on the kinetic hard-modelling of spectroscopic data

Billeter et al, Chemom. Intell. Lab. Syst., 95 (2009), 170 – 187

Kinetic hard-modelling and spectral validation
of rank-deficient spectroscopic data: a case study

Billeter et al, Chemom. Intell. Lab. Syst., submitted
